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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.002 Å R factor = 0.033 wR factor = 0.083 Data-to-parameter ratio = 11.6

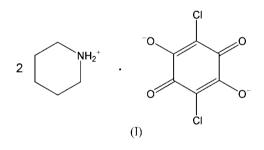
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis(piperidinium) chloranilate

In the title crystal structure, $2C_5H_{12}N^+ \cdot C_6Cl_2O_4^{2-}$, chloranilate and piperidinium ions are connected by bifurcated N— $H \cdot \cdot \cdot O$ hydrogen bonds, giving a centrosymmetric chloranirate-piperidinium 1:2 unit. The 1:2 units are connected to each other by N— $H \cdot \cdot \cdot O$ hydrogen bonds, forming a molecular ladder. There is a centre of symmetry at the centre of the anion ring.

Comment

Crystal structures in the chloranilic acid (2,5-dichloro-3,6dihydroxy-1,4-benzoquinone)-secondary amine system have been analyzed for 1:1 salts of morpholine (Ishida & Kashino, 1999), diethylamine (Ishida & Kashino, 2000), piperidine (Fukunaga & Ishida, 2003) and 1,2,3,4-tetrahydroquinoline (Ishida, 2004*b*), and for a 1:2 salt of pyrrolidine (Ishida, 2004*a*). In the present study, we have prepared the 1:2 salt (I) and determined its crystal structure at 100 K in order to extend the previous studies.



In the crystal structure of (I), the asymmetric unit is composed of $C_5H_{12}N^+ \cdot 0.5C_6O_4Cl_2^{2-}$, and an acid-base interaction involving proton transfer is observed between chloranilic acid and piperidine (Fig. 1). There is a centre of symmetry at the centre of the anion ring. The chloranilate and piperidinium ions are connected by asymmetric bifurcated

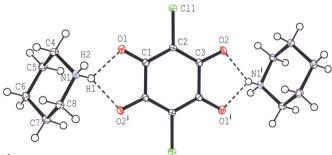


Figure 1

View of (I), with the atom labelling. Displacement ellipsoids of non-H atoms are drawn at the 50% probability level. $N-H \cdots O$ hydrogen bonds are indicated by dashed lines (the symmetry code is as in Table 1).

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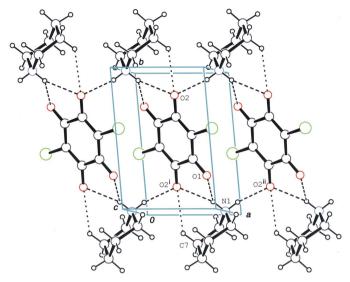


Figure 2

Packing view of (I), showing a ladder along the *a* axis. $N-H \cdots O$ and C- $H \cdots O$ hydrogen bonds are indicated by dashed and dotted lines, respectively (the symmetry codes are as in Table 2).

N-H...O hydrogen bonds (Table 2) to give a centrosymmetric chloranilate-piperidinium 1:2 unit. Within the unit, there is also a weak $C-H \cdots O$ hydrogen bond. The 1:2 units are connected by other $N-H \cdots O$ hydrogen bonds, forming a ladder running parallel to the *a* axis. The packing scheme is similar to that found in bis(pyrazolium) chloranilate (Ishida & Kashino, 2001) and bis(3-methylpyrazolium) chloranilate (Ishida, 2004c), but quite different from that in bis-(pyrrolidinium) chloranilate (Ishida, 2004a), where the two cations and one anion are arranged in an alternating manner to form a tape.

Experimental

To a solution of piperidine (85 mg, 1 mmol) in CH₃OH (10 ml), a solution of chloranilic acid (209 mg, 1 mmol) in CH₃OH (25 ml) was added at room temperature. The solution was allowed to evaporate slowly at room temperature. Dark-red crystals of (I) suitable for X-ray diffraction were formed, filtered off and dried under vacuum.

Crystal data

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$2C_5H_{12}N^+ \cdot C_6Cl_2O_4^{2-}$	V = 418.25 (5) Å ³
$M_r = 379.28$	Z = 1
Triclinic, $P\overline{1}$	$D_x = 1.506 \text{ Mg m}^{-3}$
a = 5.1239 (3) Å	Mo $K\alpha$ radiation
b = 8.7058 (7) Å	$\mu = 0.41 \text{ mm}^{-1}$
c = 10.4508 (7) Å	T = 100 (2) K
$\alpha = 114.412 \ (3)^{\circ}$	Prism, dark red
$\beta = 95.561 \ (2)^{\circ}$	$0.28 \times 0.15 \times 0.08 \text{ mm}$
$\gamma = 95.070 \ (3)^{\circ}$	
Data collection	

Rigaku R-AXIS RAPID II diffractometer ω scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{\min} = 0.781, \ T_{\max} = 0.968$

m

4040 measured reflections 1838 independent reflections 1534 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.045$ $\theta_{\rm max} = 27.1^\circ$

Refinement	

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0364P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.033$	+ 0.1379P]
$wR(F^2) = 0.083$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} = 0.001$
1838 reflections	$\Delta \rho_{\rm max} = 0.38 \text{ e } \text{\AA}^{-3}$
158 parameters	$\Delta \rho_{\rm min} = -0.38 \text{ e } \text{\AA}^{-3}$
All H-atom parameters refined	

Table 1 Selected bond lengths (Å).

Cl1-C2	1.7466 (16)	N1-C8	1.494 (2)
O1-C1	1.2372 (19)	C1-C2	1.423 (2)
O2-C3	1.2662 (18)	$C1-C3^{i}$	1.545 (2)
N1-C4	1.493 (2)	C2-C3	1.389 (2)

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

Table 2			
Hydrogen-bond	geometry	(Å,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
	0.88 (2)	2.13 (2)	2.8045 (19)	133.1 (17)
	0.88 (2)	2.20 (2)	2.9929 (19)	150.7 (17)
	0.86 (2)	1.98 (2)	2.8070 (19)	162 (2)
	0.99 (2)	2.59 (2)	3.349 (2)	133.5 (17)

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) -x + 2, -y + 1, -z + 1.

H atoms were located in a Fourier map and refined isotropically; N-H and C-H bond lengths are in the range 0.86 (2)-0.88 (2) Å and 0.95 (2)-1.01 (2) Å, respectively.

Data collection: PROCESS-AUTO (Rigaku/MSC, 2004); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2004); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: CrystalStructure.

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