Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Moamen S. Refat, ${ }^{\text {a }}$ Hamdy Al-Didamony Ahmed, ${ }^{\text {b }}$ Lamia A. El-Zayat, ${ }^{\text {a }}$ Takeo Fukunaga ${ }^{\mathrm{c}}$ and Hiroyuki Ishida ${ }^{\mathrm{c} *}$

${ }^{\text {a }}$ Chemistry Department, Faculty of Education, Suez Canal University, Port Said, Egypt,
${ }^{\text {b }}$ Chemistry Department, Faculty of Science,
Zagazig University, Egypt, and ${ }^{\text {c }}$ Department of Chemistry, Faculty of Science, Okayama University, Okayama 700-8530, Japan

Correspondence e-mail:
ishidah@cc.okayama-u.ac.jp

## Key indicators

Single-crystal X-ray study
$T=100 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.033$
$w R$ factor $=0.083$
Data-to-parameter ratio $=11.6$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## Bis(piperidinium) chloranilate

In the title crystal structure, $2 \mathrm{C}_{5} \mathrm{H}_{12} \mathrm{~N}^{+} \cdot \mathrm{C}_{6} \mathrm{Cl}_{2} \mathrm{O}_{4}{ }^{2-}$, chloranilate and piperidinium ions are connected by bifurcated N $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, giving a centrosymmetric chlorani-rate-piperidinium 1:2 unit. The 1:2 units are connected to each other by $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, forming a molecular ladder. There is a centre of symmetry at the centre of the anion ring.

## Comment

Crystal structures in the chloranilic acid (2,5-dichloro-3,6-dihydroxy-1,4-benzoquinone)-secondary amine system have been analyzed for $1: 1$ salts of morpholine (Ishida \& Kashino, 1999), diethylamine (Ishida \& Kashino, 2000), piperidine (Fukunaga \& Ishida, 2003) and 1,2,3,4-tetrahydroquinoline (Ishida, 2004b), and for a 1:2 salt of pyrrolidine (Ishida, 2004a). In the present study, we have prepared the 1:2 salt (I) and determined its crystal structure at 100 K in order to extend the previous studies.
2


(I)

In the crystal structure of (I), the asymmetric unit is composed of $\mathrm{C}_{5} \mathrm{H}_{12} \mathrm{~N}^{+} \cdot 0.5 \mathrm{C}_{6} \mathrm{O}_{4} \mathrm{Cl}_{2}{ }^{2-}$, and an acid-base interaction involving proton transfer is observed between chloranilic acid and piperidine (Fig. 1). There is a centre of symmetry at the centre of the anion ring. The chloranilate and piperidinium ions are connected by asymmetric bifurcated


Figure 1
View of (I), with the atom labelling. Displacement ellipsoids of non-H atoms are drawn at the $50 \%$ probability level. $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds are indicated by dashed lines (the symmetry code is as in Table 1).


Figure 2
Packing view of (I), showing a ladder along the $a$ axis. $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-$ $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds are indicated by dashed and dotted lines, respectively (the symmetry codes are as in Table 2).
$\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 2) to give a centrosymmetric chloranilate-piperidinium 1:2 unit. Within the unit, there is also a weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond. The 1:2 units are connected by other $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, forming a ladder running parallel to the $a$ axis. The packing scheme is similar to that found in bis(pyrazolium) chloranilate (Ishida \& Kashino, 2001) and bis(3-methylpyrazolium) chloranilate (Ishida, 2004c), but quite different from that in bis(pyrrolidinium) chloranilate (Ishida, 2004a), where the two cations and one anion are arranged in an alternating manner to form a tape.

## Experimental

To a solution of piperidine ( $85 \mathrm{mg}, 1 \mathrm{mmol}$ ) in $\mathrm{CH}_{3} \mathrm{OH}(10 \mathrm{ml})$, a solution of chloranilic acid ( $209 \mathrm{mg}, 1 \mathrm{mmol}$ ) in $\mathrm{CH}_{3} \mathrm{OH}(25 \mathrm{ml})$ was added at room temperature. The solution was allowed to evaporate slowly at room temperature. Dark-red crystals of (I) suitable for X-ray diffraction were formed, filtered off and dried under vacuum.

## Crystal data

$$
\begin{aligned}
& 2 \mathrm{C}_{5} \mathrm{H}_{12} \mathrm{~N}^{+} \cdot \mathrm{C}_{6} \mathrm{Cl}_{2} \mathrm{O}_{4}{ }^{2-} \\
& M_{r}=379.28 \\
& \text { Triclinic, } P \overline{1} \\
& a=5.1239(3) \AA \AA \\
& b=8.7058(7) \AA \\
& c=10.4508(7) \AA \\
& \alpha=114.42(3)^{\circ} \\
& \beta=95.561(2)^{\circ} \\
& \gamma=95.070(3)^{\circ}
\end{aligned}
$$

## Data collection

$$
\begin{aligned}
& V=418.25(5) \AA^{3} \\
& Z=1 \\
& D_{x}=1.506 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \mu=0.41 \mathrm{~mm}^{-1} \\
& T=100(2) \mathrm{K} \\
& \text { Prism, dark red } \\
& 0.28 \times 0.15 \times 0.08 \mathrm{~mm}
\end{aligned}
$$

Rigaku R-AXIS RAPID II
$\quad$ diffractometer
$\omega$ scans
Absorption correction: multi-scan
$\quad(A B S C O R ;$ Higashi, 1995)
$\quad T_{\min }=0.781, T_{\max }=0.968$

Rigaku R-AXIS RAPID II
difractometer
Absorption correction: multi-scan
$T_{\min }=0.781, T_{\max }=0.968$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0364 P)^{2}\right. \\
& \quad+0.1379 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.38 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.38 \mathrm{e}^{-3}
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.033$
$w R\left(F^{2}\right)=0.083$
$S=1.08$
1838 reflections
158 parameters
All H -atom parameters refined

Table 1
Selected bond lengths $(\AA)$.

| $\mathrm{Cl} 1-\mathrm{C} 2$ | $1.7466(16)$ | $\mathrm{N} 1-\mathrm{C} 8$ | $1.494(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 1-\mathrm{C} 1$ | $1.2372(19)$ | $\mathrm{C} 1-\mathrm{C} 2$ | $1.423(2)$ |
| $\mathrm{O} 2-\mathrm{C} 3$ | $1.2662(18)$ | $\mathrm{C} 1-\mathrm{C} 3^{\mathrm{i}}$ | $1.545(2)$ |
| $\mathrm{N} 1-\mathrm{C} 4$ | $1.493(2)$ | $\mathrm{C} 2-\mathrm{C} 3$ | $1.389(2)$ |

Symmetry code: (i) $-x+1,-y+1,-z+1$.

Table 2
Hydrogen-bond geometry ( $\AA^{\circ},{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 \cdots \mathrm{O} 1$ | $0.88(2)$ | $2.13(2)$ | $2.8045(19)$ | $133.1(17)$ |
| $\mathrm{N} 1-\mathrm{H} 1 \cdots \mathrm{O}^{\mathrm{i}}$ | $0.88(2)$ | $2.20(2)$ | $2.9929(19)$ | $150.7(17)$ |
| $\mathrm{N} 1-\mathrm{H} 2 \cdots \mathrm{O}^{2 i}$ | $0.86(2)$ | $1.98(2)$ | $2.8070(19)$ | $162(2)$ |
| $\mathrm{C}^{\mathrm{H}}-\mathrm{H} 9 \cdots \mathrm{O}^{2}$ | $0.99(2)$ | $2.59(2)$ | $3.349(2)$ | $133.5(17)$ |

Symmetry codes: (i) $-x+1,-y+1,-z+1$; (ii) $-x+2,-y+1,-z+1$.

H atoms were located in a Fourier map and refined isotropically; $\mathrm{N}-\mathrm{H}$ and $\mathrm{C}-\mathrm{H}$ bond lengths are in the range $0.86(2)-0.88$ (2) $\AA$ and 0.95 (2)-1.01 (2) $\AA$, respectively.

Data collection: PROCESS-AUTO (Rigaku/MSC, 2004); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/MSC, 2004); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: CrystalStructure.

This work was partly supported by a Grant-in-Aid for Scientific Research (C) (No. 16550014) from the Ministry of Education, Science, Sports and Culture of Japan.

## References

Altomare, A., Cascarano, G., Giacovazzo, G., Guagliardi, A., Burla, M. C., Polidori, G. \& Camalli, M. (1994). J. Appl. Cryst. 27, 435.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Fukunaga, T. \& Ishida, H. (2003). Acta Cryst. E59, o1793-o1795.
Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
Ishida, H. (2004a). Acta Cryst. E60, o974-o976.
Ishida, H. (2004b). Acta Cryst. E60, o1674-o1676.
Ishida, H. (2004c). Acta Cryst. E60, o2506-o2508.
Ishida, H. \& Kashino, S. (1999). Acta Cryst. C55, 1923-1926.
Ishida, H. \& Kashino, S. (2000). Acta Cryst. C56, e202-e204.
Ishida, H. \& Kashino, S. (2001). Acta Cryst. C57, 476-479.
Rigaku/MSC (2004). PROCESS-AUTO and CrystalStructure (Version 3.7.0). Rigaku/MSC, The Woodlands, Texas, USA.
Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.


[^0]:    (C) 2006 International Union of Crystallography All rights reserved

